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Национален план за
възстановяване и устойчивост



НА РЕПУБЛИКА БЪЛГАРИЯ

Zeolitic materials for new applications

Valentin Valtchev

До говор BG-RRP-2.004-0008 за финансиране на проект „СОФИЙСКИ УНИВЕРСИТЕТ – МАРКЕРА ЗА ИНОВАЦИИ И ТЕХНОЛОГИЧЕН ТРАНСФЕР (SUMMIT)“ по статъч 2 „Създаване на мрежа от изследователски висши училища“ в рамките на компонента „Иновативна България“ от Национален план за възстановяване и устойчивост към програмата за ускоряване на икономическия растеж в съответствие с инициативата за трансформация чрез наука и иновации



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МАРКЕР ЗА ИНОВАЦИИ И ТЕХНОЛОГИЧЕН ТРАНСФЕР



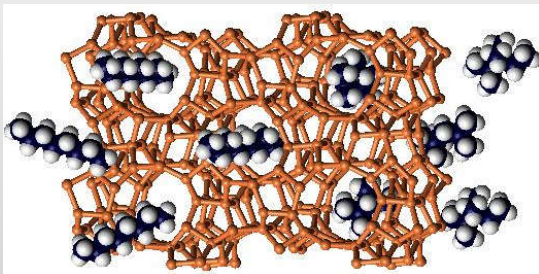
Defintion: Zeolites are crystalline aluminosilicates, composed of TO_4 tetrahedra ($T = Si, Al$) with O atoms connecting neighbouring tetrahedral, that contain pores and cavities of molecular dimensions (Breck, 1974).

Properties

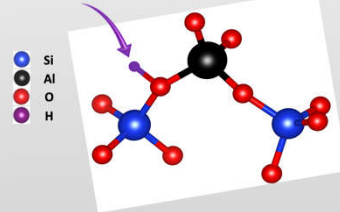
- Tunable acidity
- High surface area
- High thermal stability
- High chemical stability
- Well-defined micro-pore structure
- Shape-selectivity

Applications

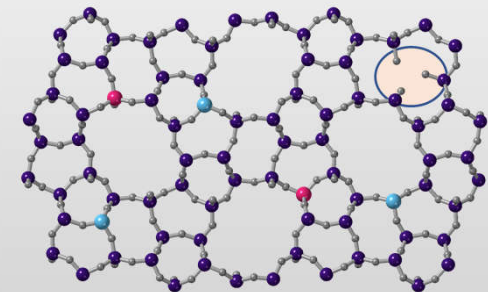
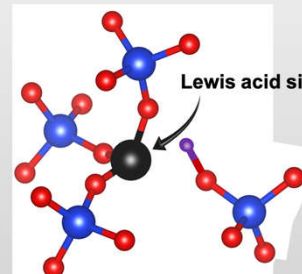
- Heterogeneous catalysts in oil refining, petrochemical industry, fine chemicals
- Molecular sieves
- Separation of toxic gases and radioactive isotopes
- Environmental applications



Brønsted acid site



Lewis acid site





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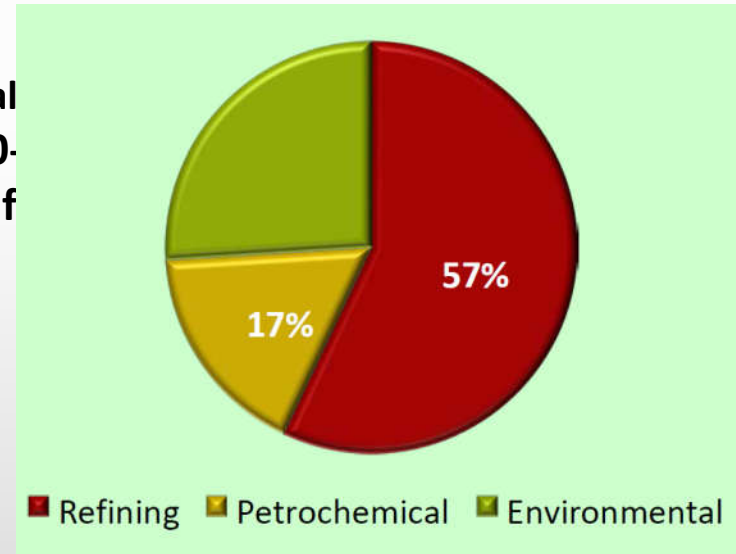
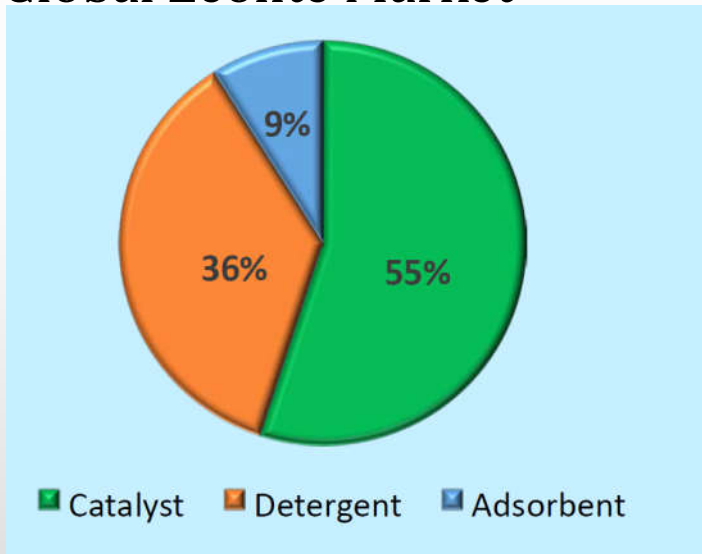
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Michel BOUDART: «Catalysis Blends Elegance and Relevance»

Global Zeolite Market



Refining: FCC, Hydrocracking, Dewaxing Lubes/Distillates, Isomerization

Petrochemical: Xylenes (p-Xylene), Cumene, Ethyl Benzene

Environmental: Non-stationary and stationary



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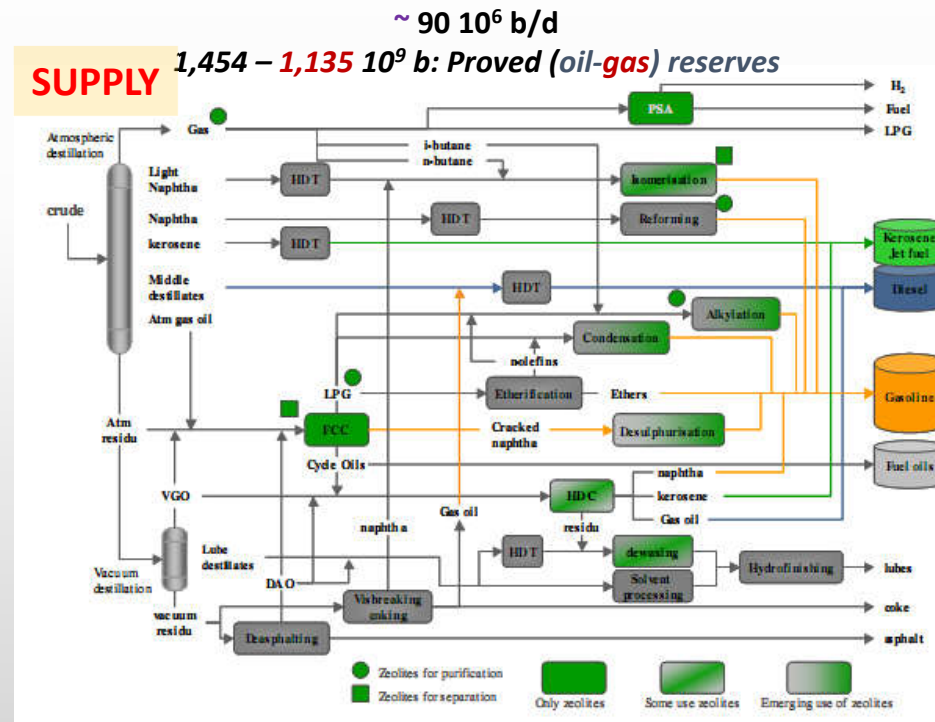
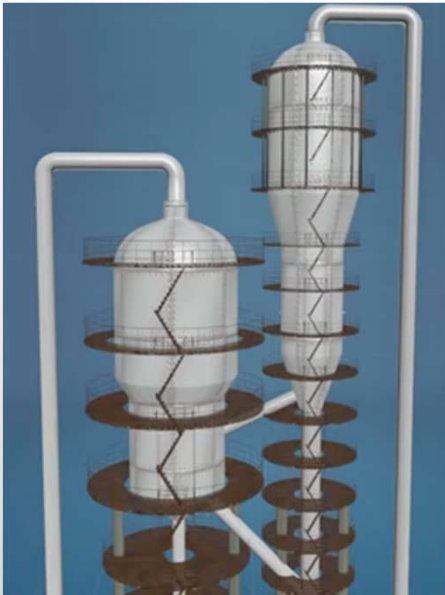


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Zeolites = Game Changer in FCC (60's)

Introduction of zeolites in FCC (1964) – a Disruptive Technology

Fluid Catalytic Cracking (FCC) a milestone in catalysis



- Changes in Transportation Fuels
 - Gasoline / Diesel shift
 - Higher quality, higher octane gasoline
- Chemicals (aromatics and light olefins) and high performance lubricants demand are up
- Changes in Feedstock
 - Shale Gas / Tight Oil
 - Heavy Oils
 - Renewables enter feed mix, but slowly
- Environmental Needs
 - Fuels regulations and emission standards
- Energy Efficiency

Energy transition – the big challenge of nowadays

New materials for the current/future needs of chemical process industries

- Oil-to-chemicals
- Methane-to-chemicals
- Methane as a transition fuel
- Renewable feedstocks
- Plastics processing
- Greenhouse gasses separation/processing
- New energy vectors

What do we need?

- Hydrothermally stable
- Ultra hydrophobic
- Ultra thermally stable
- Ultra-large pore zeolites
- Single acid site catalysis
- Improved accessibility



Going beyond the conventional zeolite comfort zone

Extending the potential applications of zeolites – thinking about tomorrow

1. Medical applications
2. Drinkable water production from air humidity
3. New materials for energy storage
4. Wastes processing

How to get what we need?

Two complementary approaches to design zeolites



Zeolite Crystal Engineering
Design “fit-for-purpose”
materials from existing
structures (particle size &
shape, composition, added
meso--& macro-porosity,
defects control, Al siting...).

FAU, MOR, CHA, FER, LTL
MFI, BEA*, LTA

ABW	ACO	AEI	AEL	AEN	AET	AFG	AFI	AFN	AFO	AFR	AFS	AFT	AFV	AFX
AFY	AHT	ANA	APC	APD	AST	ASV	ATN	ATO	ATS	ATT	ATV	AVL	AWO	AWW
BCT	BEC	BIK	BOF	BOG	BOZ	BPH	BRE	BSV	CAN	CAS	CDO	CFI	CGF	CGS
CHA	-CHI	-CLO	CON	CSV	CZP	DAC	DDR	DFO	DFT	DOH	DON	EAB	EDI	EEL
EMT	EON	EPI	ERI	ESV	ETL	ETR	EUO	EZT	FAR	FAU	FER	FRA	GIS	GIU
GME	GON	GOO	HEU	IFO	IFR	-IFU	IFW	IFY	IHW	IMF	IRN	IRR	-IRY	ISV
ITE	ITG	ITH	ITR	ITT	-ITV	ITW	IWR	IWS	IWV	IWW	JBW	JNT	JOZ	JRY
JSN	JSR	JST	JSW	KFI	LAU	LEV	LIO	-LIT	LOS	LOV	LTA	LTF	LTJ	LTL
LTN	MAR	MAZ	MEI	MEL	MEP	MER	MFI	MFS	MON	MOR	MOZ	MSE	MSO	MTF
MTN	MTT	MTW	MVY	MWF	MWW	NAB	NAT	NES	NON	NPO	NPT	NSI	OBW	OFF
OKO	OSI	OSO	OWE	-PAR	PAU	PCR	PHI	PON	POS	PSI	PUN	RHO	-RON	RRO
RSN	RTE	RTH	RUT	RWR	RWY	SAF	SAO	SAS	SAT	SAV	SBE	SBN	SBS	SBT
SEW	SFE	SFF	SFG	SFH	SFN	SFO	SFS	SFW	SGT	SIV	SOD	SOF	SOS	SSF
SSY	STF	STI	STT	STW	-SVR	SVV	SZR	TER	THO	TOL	TON	TSC	TUN	UEI
UFI	UOS	UOV	UOZ	USI	UTL	UWY	VET	VFI	VNI	VSV	WEI	-WEN	YUG	ZON

Find a New Framework Type
You may hit the jackpot and
discover a new material with
unique properties.

MWW, TON, MTT,
MFI, BEA*, LTA



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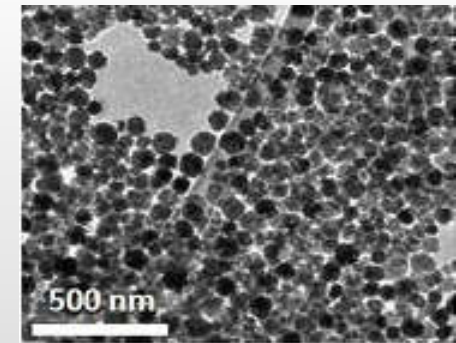
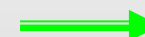
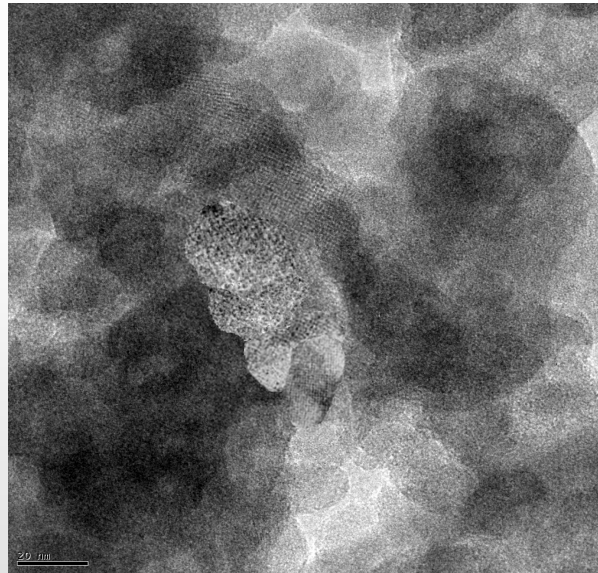
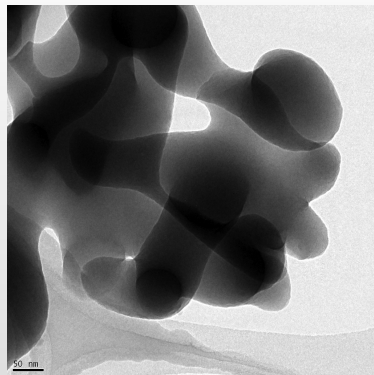
Our approach: Zeolite Crystal Engineering

- Customized zeolite synthesis
 - Control of zeolite crystal size and morphology
- Post-synthesis modification of zeolites properties
 - Modification of zeolite framework composition
 - Generation of mesopore to improve the intercrystalline diffusion
- Combining customized zeolite synthesis with post-synthesis modification to obtain the desired material
- Development of new synthetic methods



In-situ control of zeolite crystal size

- Which stage of gel evolution are the viable nuclei formed at?
- What is the spatial and temporal location of the nucleation events?

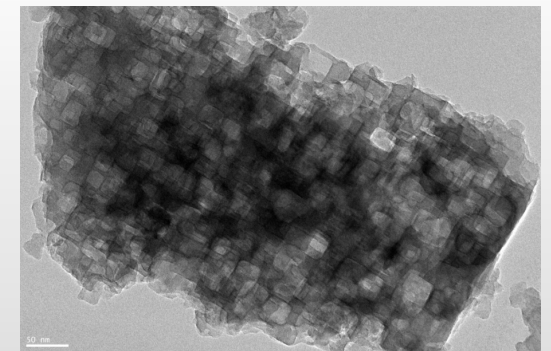
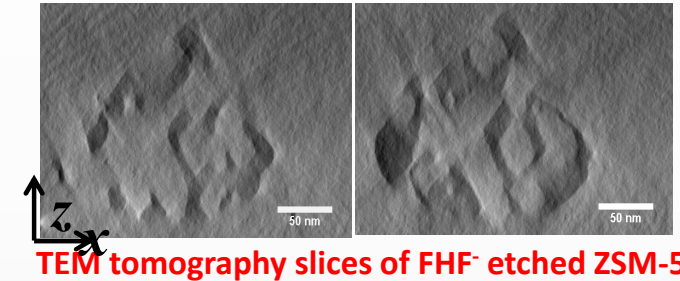
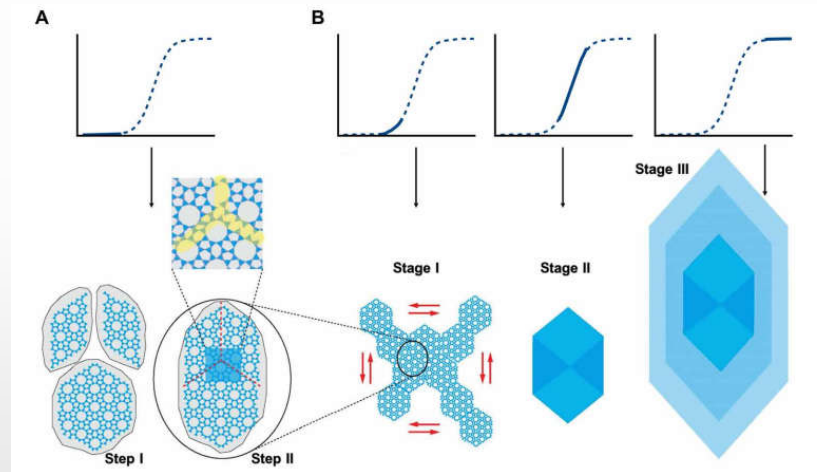
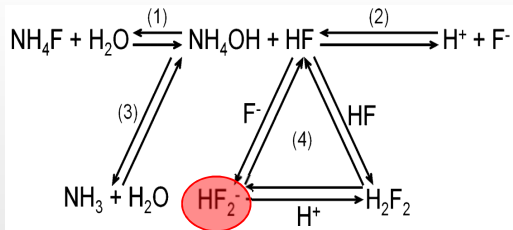
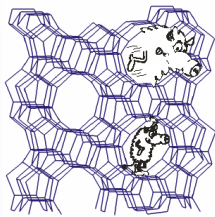


The zeolite crystal size determines:

- The diffusion pathway
 - The accessibility to the active sites
 - The adsorption/desorption kinetics
- The external surface area

Science, 1999, 238, 958; *J. Phys. Chem. B* 2004, 108, 15587; *J. Am. Chem. Soc.* 2005, 127, 16171; *Langmuir* 2005, 21, 10724; *Micropor. Mesopor. Mater.* 2007, 101, 73; *J. Am. Chem. Soc.* 2009, 131, 10127; *Chem. – A Eur. J.* 2011, 17, 2199; *Chem. – A Eur. J.* 2015, 21, 18316; *J. Am. Chem. Soc.* 2021, 143, 1993.

Post-synthesis control of active sites accessibility in zeolites



Open the route to:

- Decrease the impact of the diffusion limitations
- Healing the defect sites
- Incorporation of catalytically active species
- Retaining only the coherent domains in a zeolite crystal

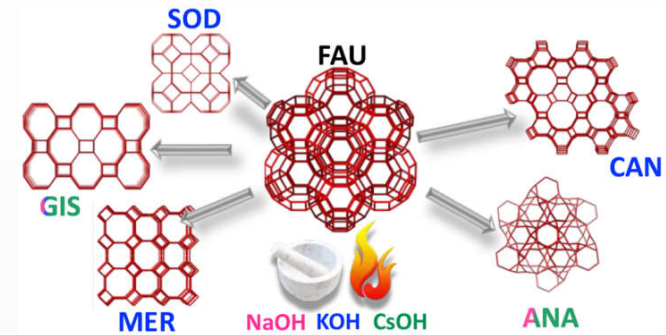
Advanced Science 2022, 9, 2104414; Advanced Science 2021, 2100001; Science
 Advances 2021, 7, eabg0454; J. Am. Chem. Soc. 2017, 139 (48), 17273-17276;
 Angew. Chem. 2016, 55, 15049-15052; Chem. Mater. 2013, 25, 2759-2766.

First results

Title: Mechanochemically Induced OSDA-Free Interzeolite Conversion

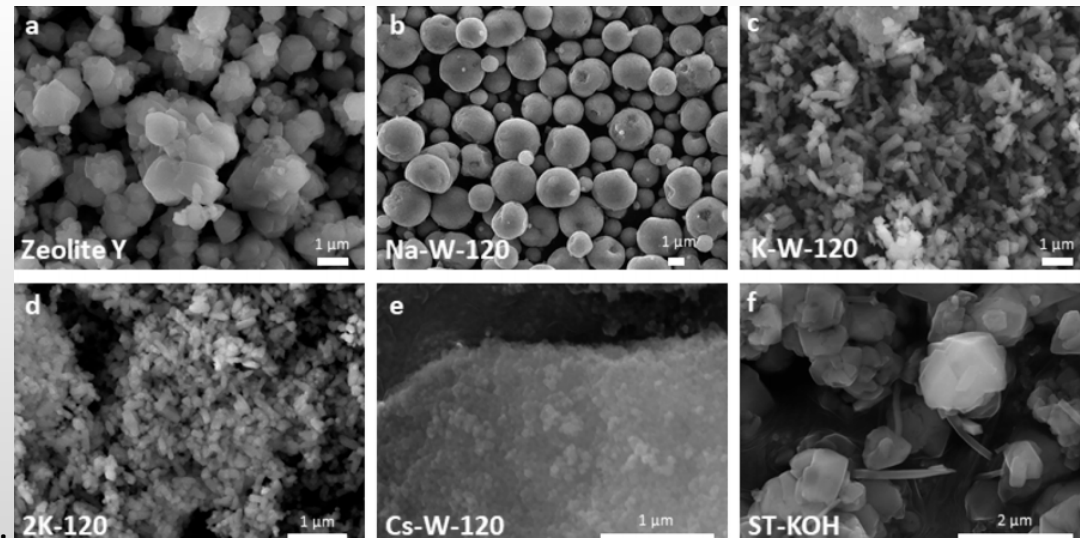
ACS Sustainable Chemical Engineering, 2024, accepted

DOI: <https://doi.org/10.1021/acssuschemeng.3c08477>



Summary

- Mechanochemical processing under controlled elevated temperatures allowed for rapid, efficient, and controllable solid-state interzeolite conversion without added OSDAs.
- Milling FAU-type zeolite Y with alkali hydroxides at 110 °C afforded three phase-pure crystalline products (ANA, MER, and CAN), mixtures of several zeolite phases, and an amorphous phase that was detected in some products.
- Our future work will focus on developing strategies to achieve the direct formation of advanced zeolite materials by mechanochemical processing and further our efforts presented herein to achieve selective interzeolite conversions.





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Thank you for your attention



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